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Impact of Oxygenates in Diesel Fuel Blends on Engine Emissions and Combustion Properties

The logo for the Institute for Powertrains & Automotive Technology (IFAO) features the letters 'IFAO' in a bold, blue, sans-serif font. A blue swoosh underline starts under the 'O' and curves around to underline the 'A'.

Institute for Powertrains
& Automotive Technology



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Impact of Oxygenates in Diesel Fuel Blends on Engine Emissions and Combustion Properties

Outline

- Introduction and Task Description
- Technical Data of Test Engine
- Chemical Analysis
- Combustion Properties
- Consumption
- Engine Emissions
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Introduction

- Climate protection conventions (e.g. Kyoto Protocol)
 - ➔ Increased utilization of renewable primary products

- EU: Renewable Energy Directive (RED) and Fuels Quality Directive (FQD)
 - ➔ 10 energ.% biofuels from transport fuel pool
 - ➔ 20% green house gas savings

- Currently known substitutes: FAME, BTL or HVO
 - ➔ Bad quality (viscosity, oxidation stability) for higher blends (FAME)
 - ➔ Inadequate raw material base (BTL, HVO)
 - ➔ Expensive production process (BTL , HVO)
 - ➔ Maximum 7 vol.% FAME approved by car manufacturers

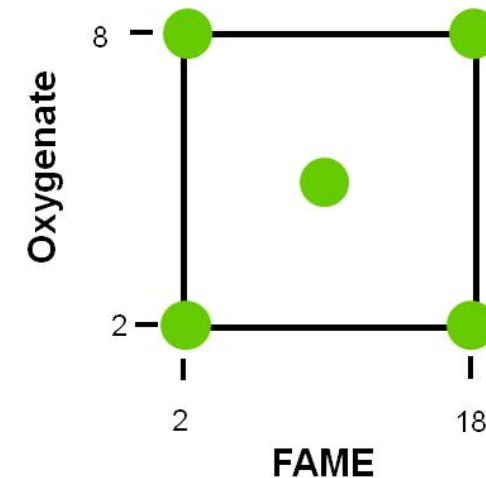
Task Description

- Investigation of new biogenic oxygenates which fulfill ethical, ecological and economical requirements
 - ➔ Diesel substitute
 - ➔ Interaction with FAME containing diesel

- Investigated oxygenates
 - ➔ Glyme, Alcohol, Polyether, Tributylcitrat, Levulinat, Valeriat

- Design of Experiments of fuel blends

- Presented oxygenates
 - ➔ Glyme (Tetra-Glyme)
 - ➔ Alcohol (Butanol)



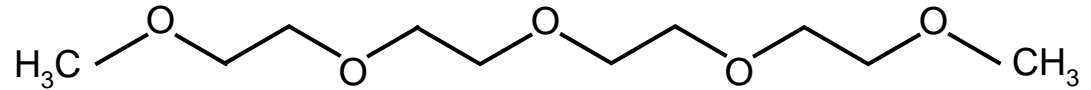
Technical Data of Test Engine

Diesel engine N47 D20 U1 (BMW)	
Cylinder	4 inline
Bore [mm]	84
Stroke [mm]	90
Displacement [cm ³]	1995
Compression Ratio	16
Power [kW]	105 at 4000 [rpm]
Max. Torque [Nm]	320 at 1750 [rpm]
Injection System	Common Rail
Turbo Charging	Var. Turbine Geometry

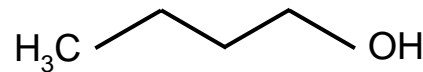


Chemical Analysis

Tetra-Glyme (C₁₀H₂₂O₅)



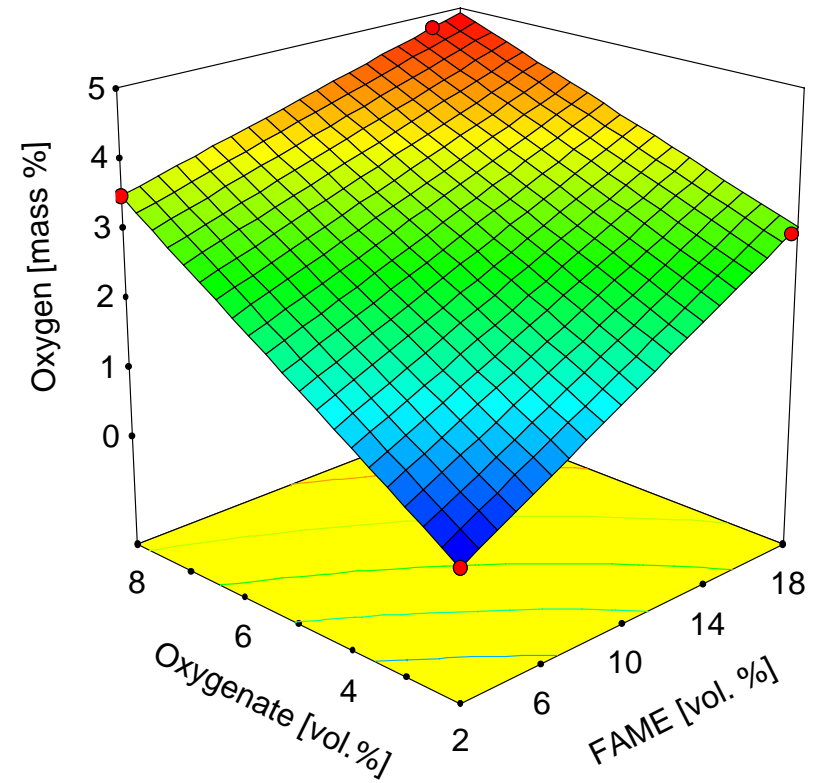
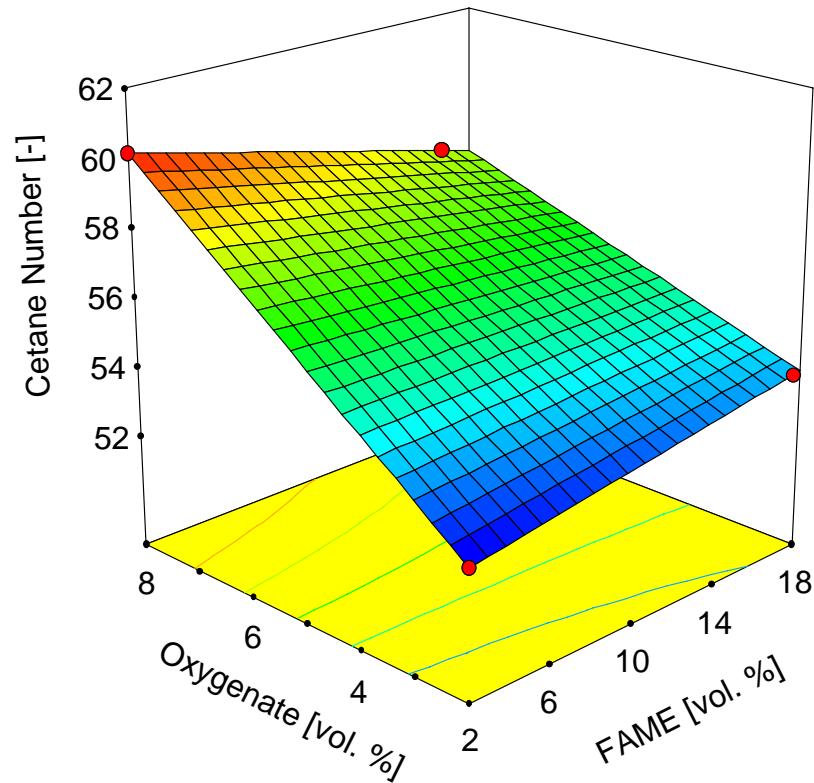
Butanol (C₄H₁₀O)



		Oxigenat	FAME	B0	C	H	O	C/H	CN	Density	Flash Point	Heat Value
		[vol. %]			[mass %]			[-]	[-]	[kg/m ³]	[°C]	[MJ/kg]
B0	-	0	0	100	86	14	0	6,14	52,1	829	61	43,5
B18	-	0	18	82	84,5	13,5	1,8	6,26	54,1	838,6	64,5	41,955
Glyme	Tetra-Glyme	8	2	90	82,9	13,4	3,5	6,19	60,2	842,9	63,5	41,209
		8	16,56	75,44	81,6	13,1	4,8	6,23	57,8	851	66,5	40,431
Alcohol	Butanol	8	2	90	84,7	13,7	1,6	6,18	48,1	828,2	39,5	42,201
		8	16,56	75,44	82,7	13,5	3,2	6,13	49,9	835,8	38,5	41,083

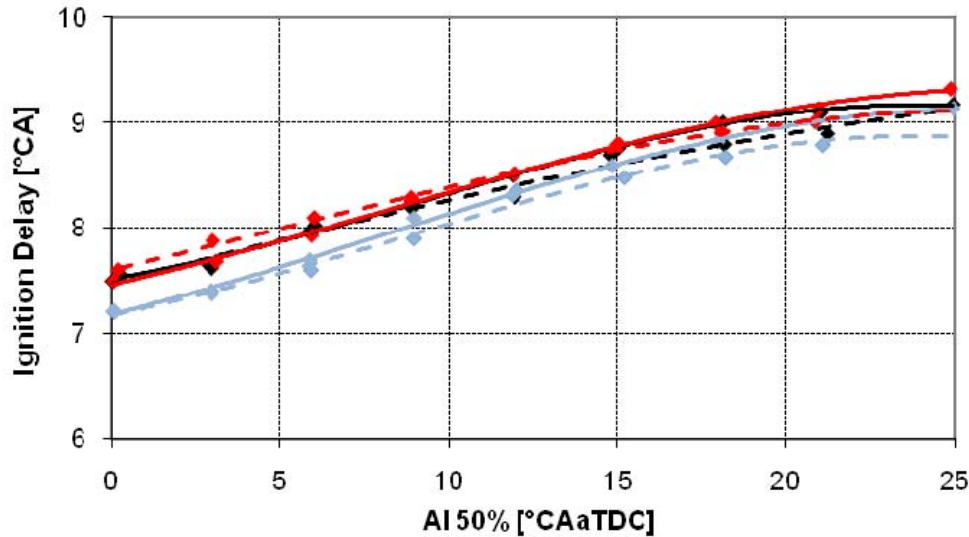
Chemical Analysis

Tetra-Glyme

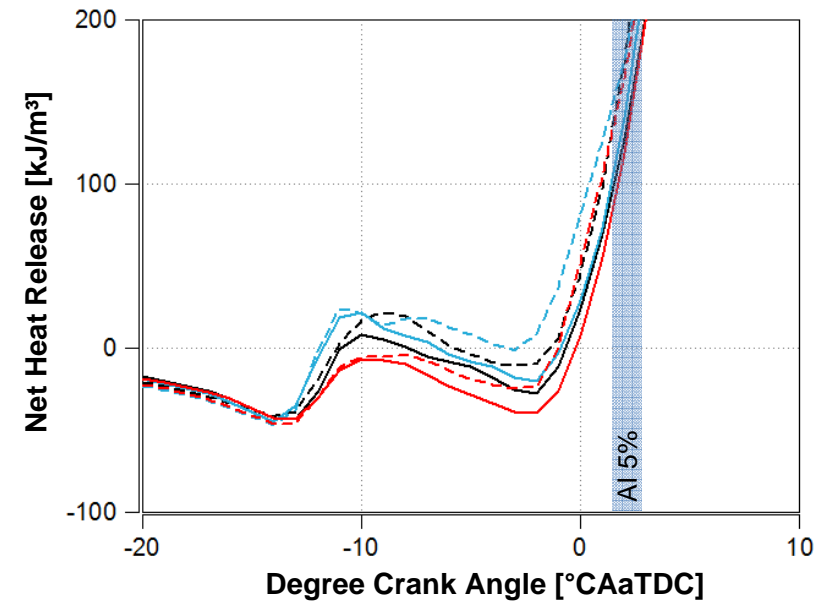


Combustion Properties

Variation of AI 50% at 2000 rpm and 15 bar BMEP



Loadpoint: 2000 rpm, 15 bar BMEP, 12°CAaTDC

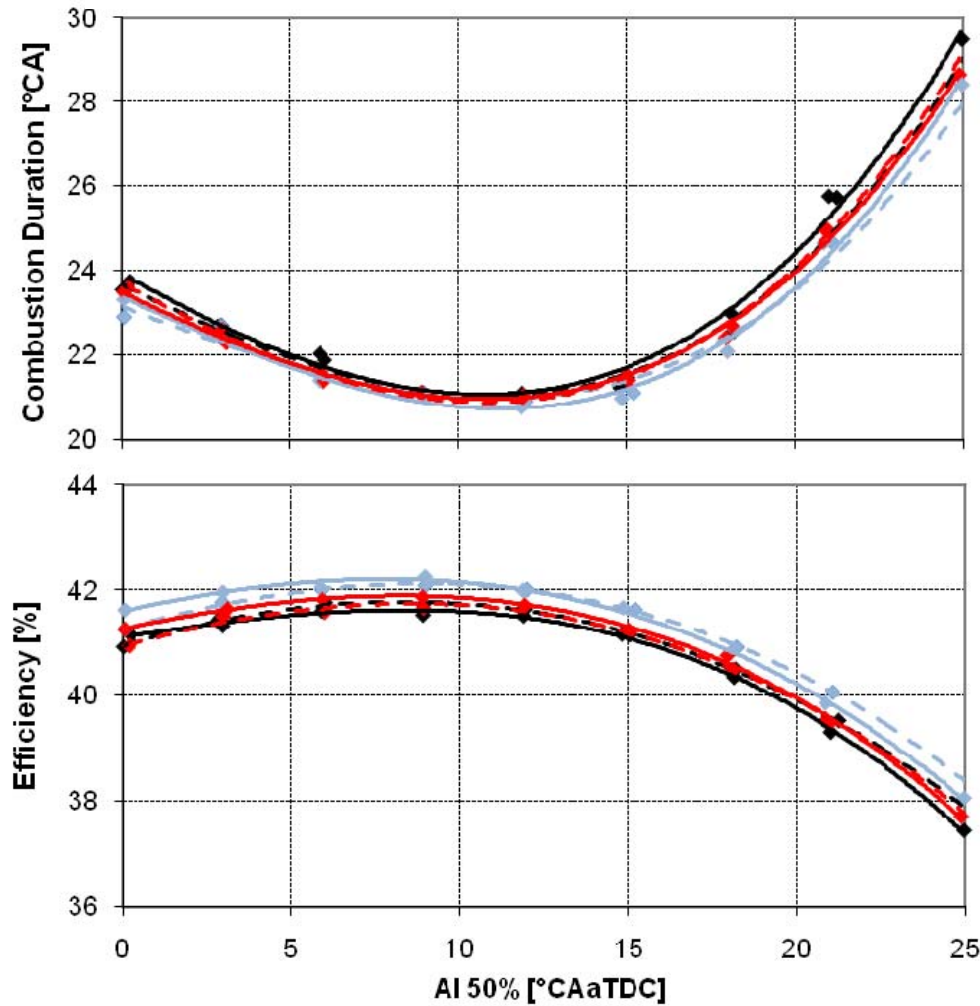


- Additivation of Tetra-Glyme increases CN
 - ➡ higher and faster net heat release
- Additivation of Butanol decreases CN
 - ➡ lower and slower net heat release
- Additivation of FAME has only minor impact on CN and combustion properties

		CN
		[-]
—	B0	52,1
- - -	B18	54,1
—	Tetra-Glyme	60,2
- - -	Tetra-Glyme + FAME	57,8
—	Butanol	48,1
- - -	Butanol + FAME	49,9

Combustion Properties

Variation of AI 50% at 2000 rpm and 15 bar BMEP

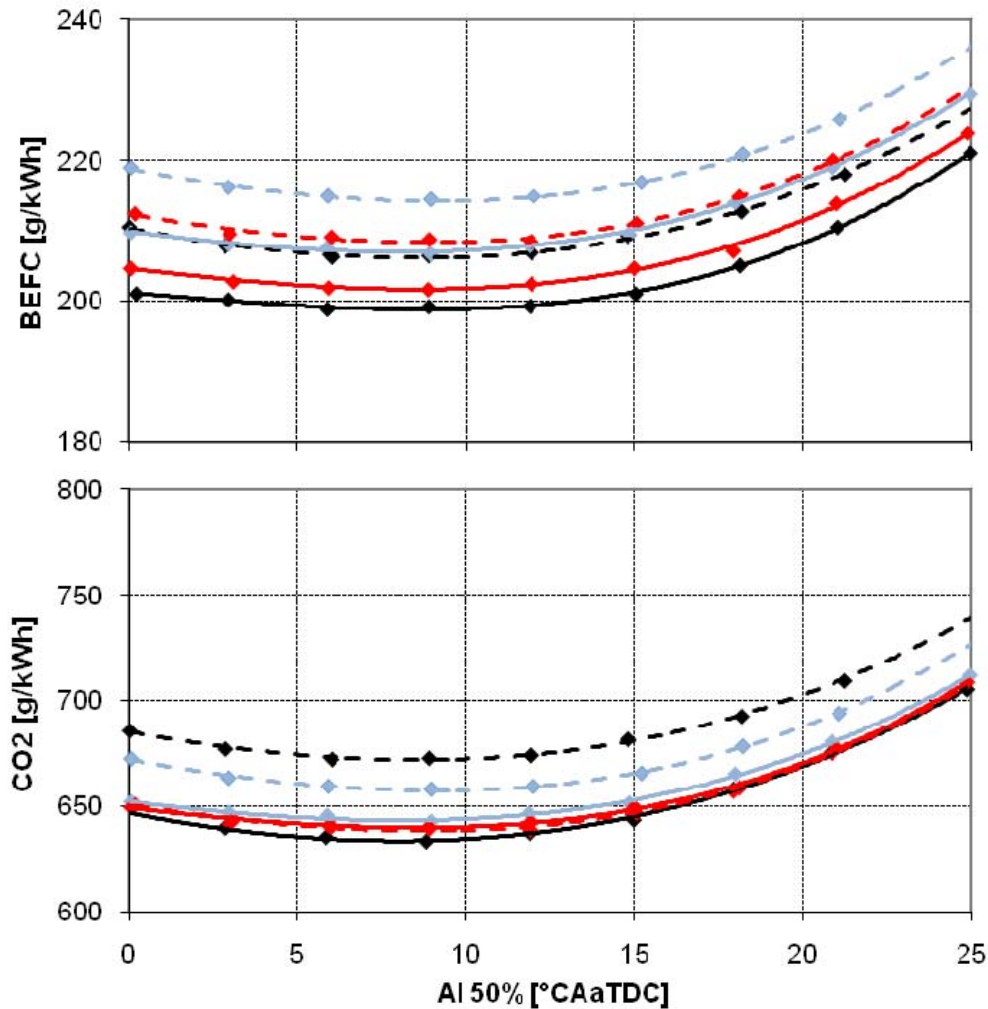


		CN	O
		[-]	[mass %]
—	B0	52,1	0
- - -	B18	54,1	1,8
—	Tetra-Glyme	60,2	3,5
- - -	Tetra-Glyme + FAME	57,8	4,8
—	Butanol	48,1	1,6
- - -	Butanol + FAME	49,9	3,2

- Additivation of Tetra-Glyme reduces combustion duration
 - ➡ Increase of efficiency
- Additivation of butanol has only minor impact on combustion duration and efficiency
- Additivation of FAME has only minor impact on combustion duration and efficiency

Consumption

Variation of AI 50% at 2000 rpm and 15 bar BMEP



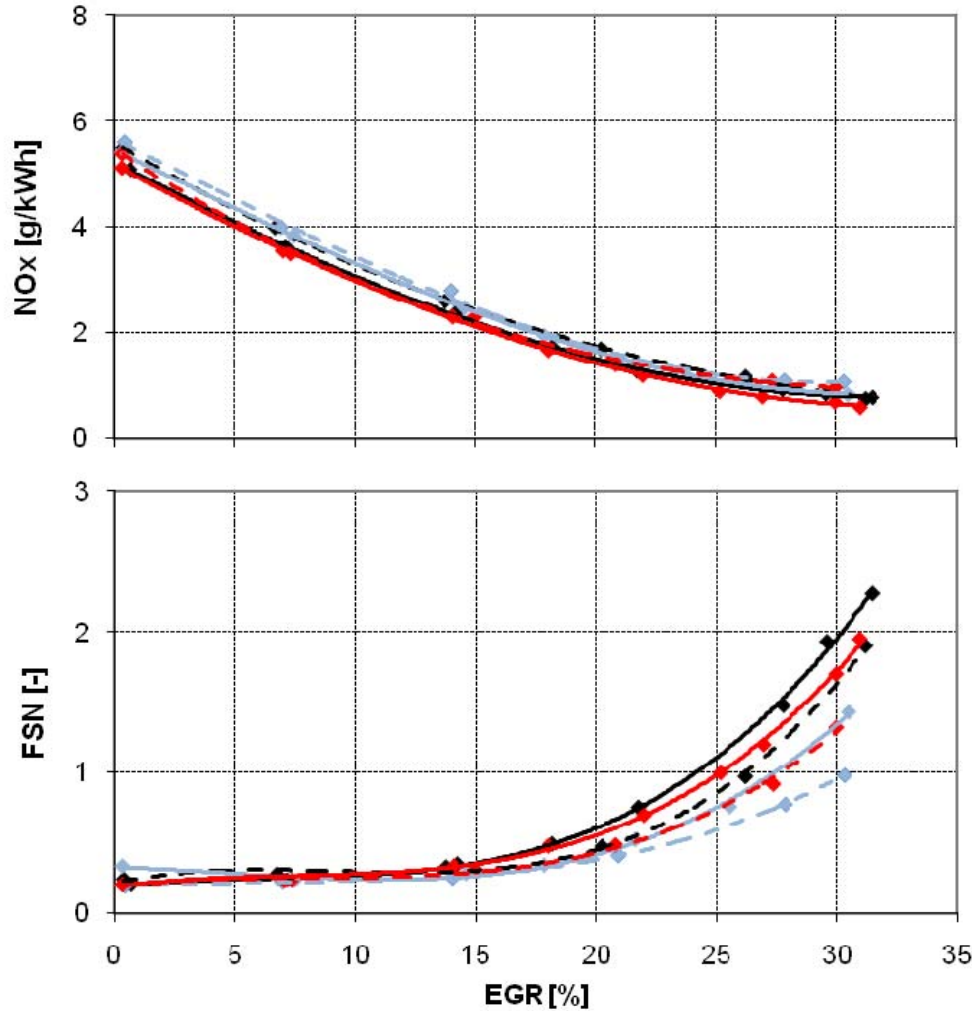
		Density	Heat Value	C
		[kg/m ³]	[MJ/kg]	[mass %]
—	B0	829	43,5	86
- - -	B18	838,6	41,955	84,5
—	Tetra-Glyme	842,9	41,209	82,9
- - -	Tetra-Glyme + FAME	851	40,431	81,6
—	Butanol	828,2	42,201	84,7
- - -	Butanol + FAME	835,8	41,083	82,7

- Additivation of oxygenate decreases heat value
 - ➔ Increase of gravimetric consumption
- Additivation of FAME further decreases heat value
 - ➔ Further increase of gravimetric consumption
- CO₂-emissions dependent on relation between gravimetric consumption and mass of carbon in fuel blend

Limited Emissions (NO_x and Particle)

		CN	O
		[-]	[mass %]
—	B0	52,1	0
- - -	B18	54,1	1,8
—	Tetra-Glyme	60,2	3,5
- - -	Tetra-Glyme + FAME	57,8	4,8
—	Butanol	48,1	1,6
- - -	Butanol + FAME	49,9	3,2

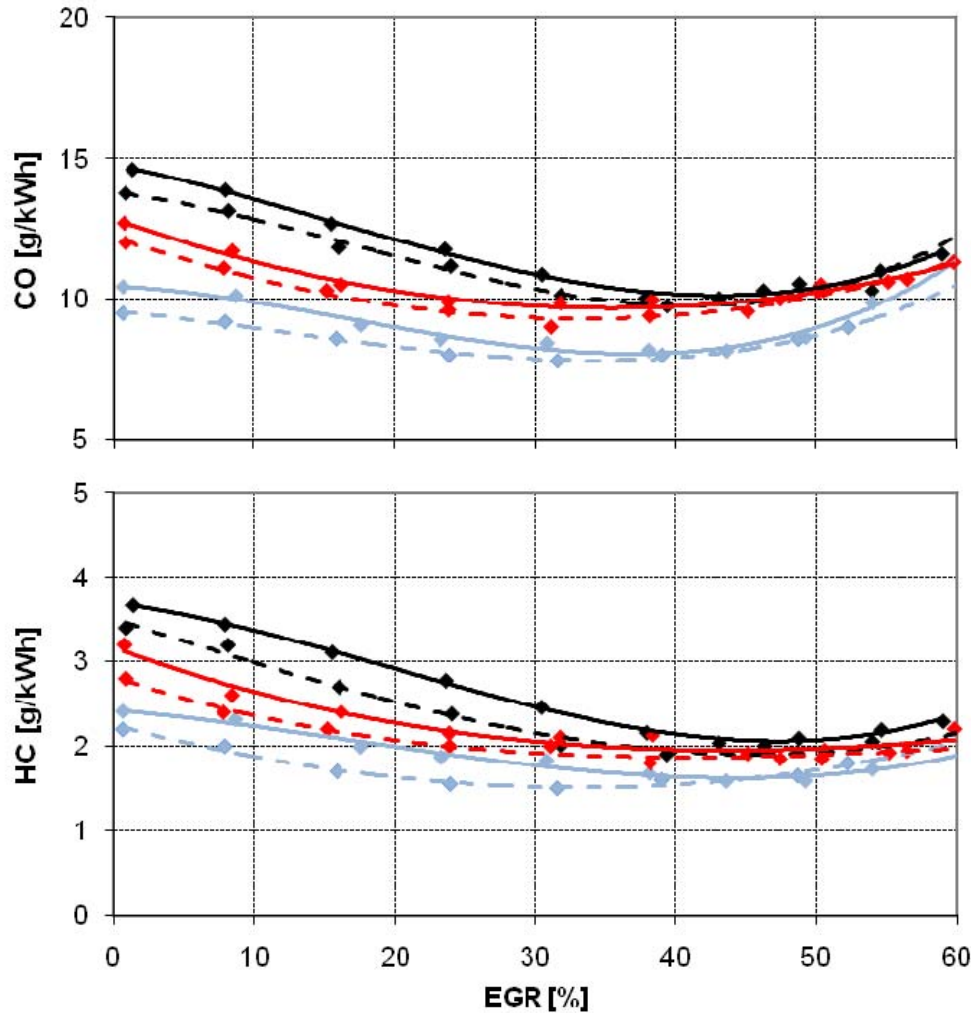
Variation of EGR at 2000 rpm and 1 bar BMEP



- Additivation of Tetra-Glyme decreases combustion duration
 - ➡ Higher combustion chamber temperatures
 - ➡ Higher NO_x-emissions
- Additivation of oxygenate and FAME increases oxygen content
 - ➡ Reduction of substoichiometric zones
 - ➡ Decrease of particle emissions

Limited Emissions (CO and HC)

Variation of EGR at 1500 rpm and 6,5 bar BMEP



		O
		[mass %]
—	B0	0
- - -	B18	1,8
—	Tetra-Glyme	3,5
- - -	Tetra-Glyme + FAME	4,8
—	Butanol	1,6
- - -	Butanol + FAME	3,2

- Additivation of oxygenate and FAME increases oxygen content
 - ➡ More homogeneous mixture preparation
 - ➡ Decrease of CO- and HC-emissions

Summary and Outlook

- The additivation of the investigated oxigenates to diesel fuel leads to:
 - ➔ a slight decrease of the combustion duration (Tetra-Glyme)
 - ➔ an increase of the gravimetric fuel consumption
 - ➔ a high decrease of the CO-, HC- and particle emissions

- The additivation of FAME to the investigated diesel-oxigenate blends:
 - ➔ has only minor impact on the combustion properties
 - ➔ increases further the gravimetric fuel consumption
 - ➔ leads to a further decrease of the CO-, HC- and particle emissions

- Biogenic oxygenates deliver an ecological alternative as a diesel substitute and enable a high decrease of the CO-, HC- and particle emissions in combination with FAME. In this context further investigations (Tributylcitrat, Valeriat, Levulinat) will be undertaken.

Thank you for your attention!



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