

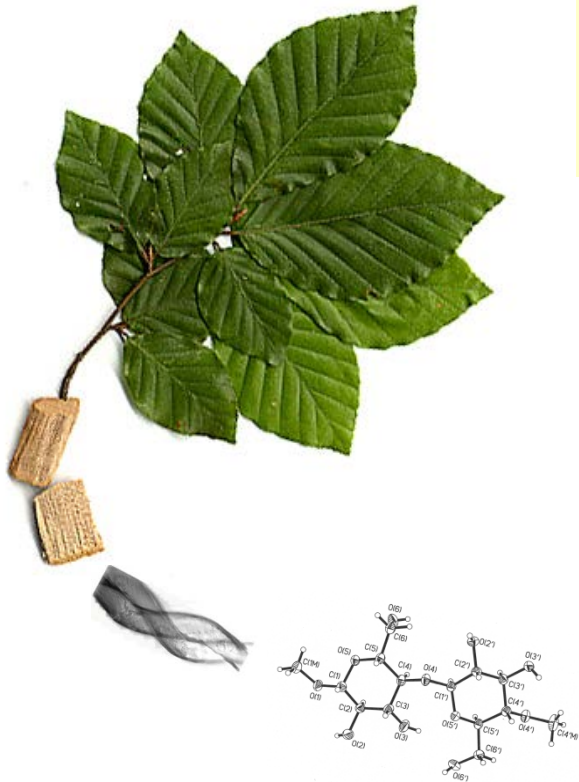
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# **Wood as the prime starting material in biorefineries**

# Wood as starting material in biorefineries



Wood and its constituents -  
in nature

Wood and its constituents -  
in biorefineries

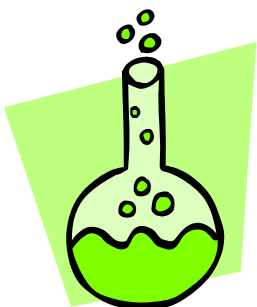
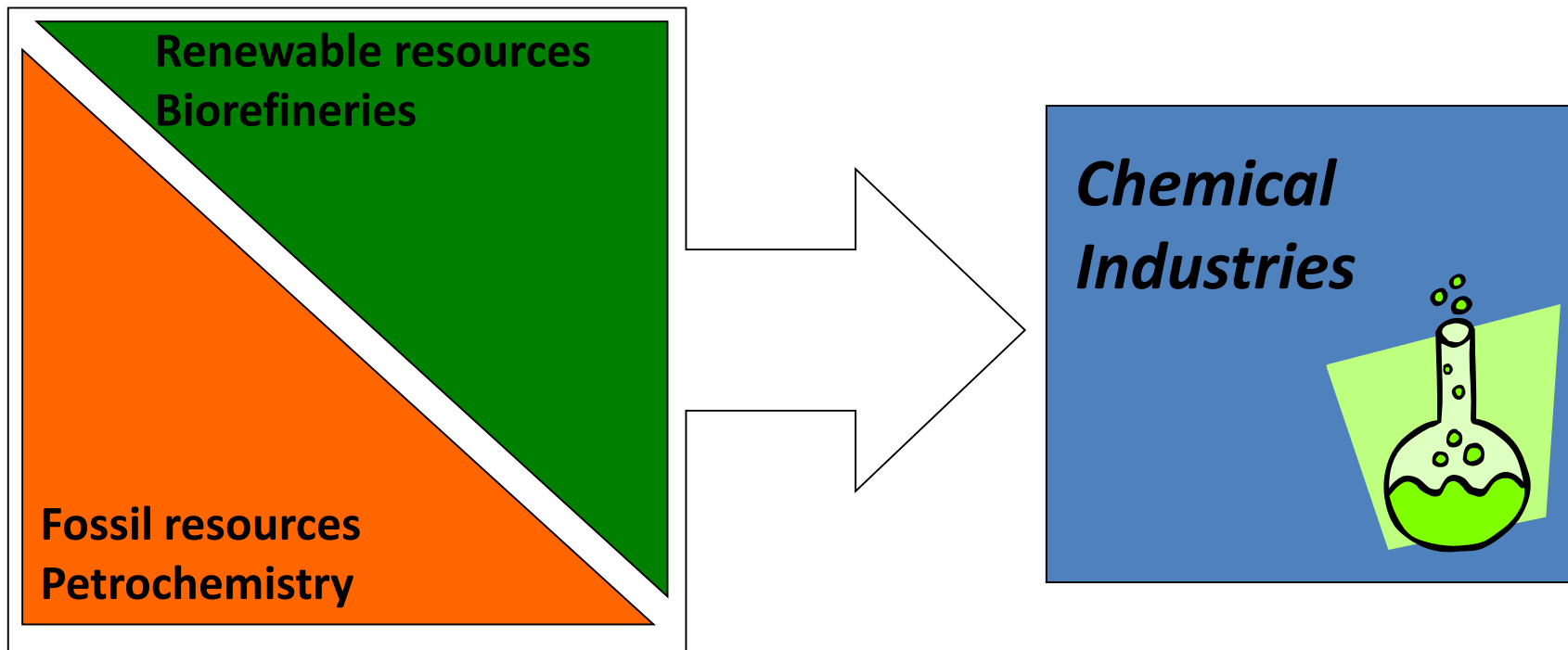
Six problems of today's  
biorefineries and green chemistry

Brief conclusion



# Looking into the (far) future

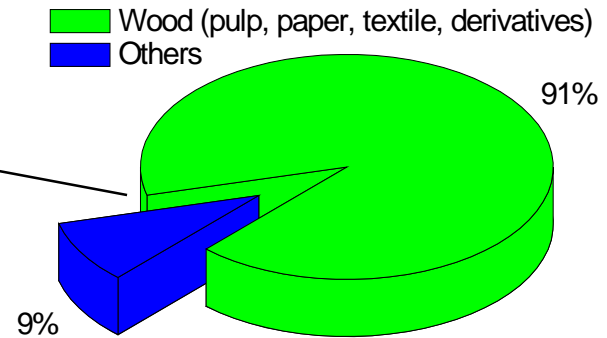
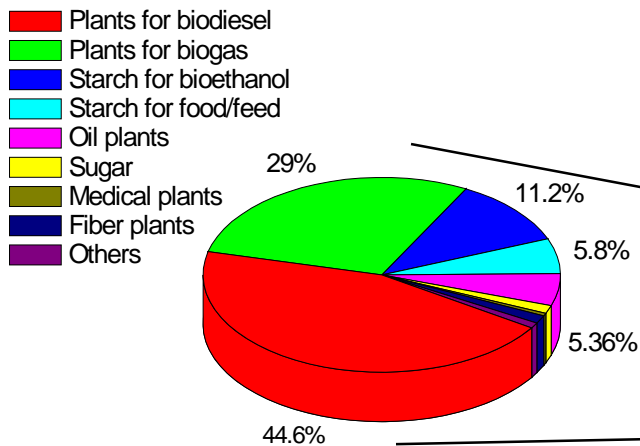
The basis of the chemical industries, present and future



In (far) future, fossil resources WILL be used up.  
If mankind is not to fall back into a rudimental, pre-industrial state, the whole production and all flows of the chemical industries will have to be changed from a **petrochemical basis** to a **renewable basis**.  
This requires long-term efforts and fundamental research.

# „Green“ starting materials for biorefineries today

## Mass balance - current situation



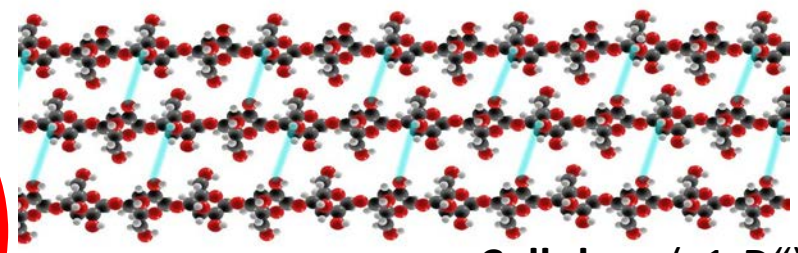
Wood is - and will remain - the most important renewable starting material for future chemical industries („biorefineries“).

**Anaximenes (ca. 550 v. Chr.):**

... and everything is composed of four elements [fire – water – earth – air], pure things out of one, most out of two or three, but wood requires all four of them...

**ὕλη – Holz, Materie**

**→ Methyl, Ethyl, Propyl**

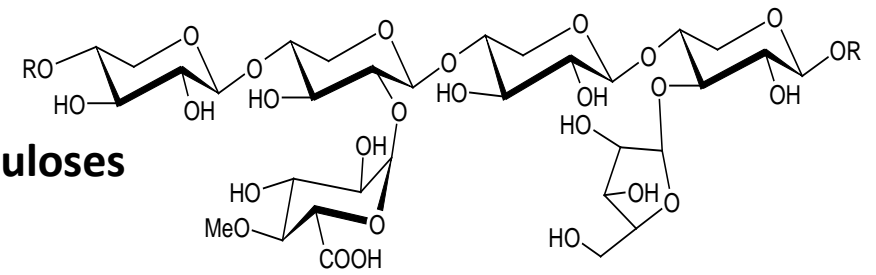


Cellulose („1-D“)

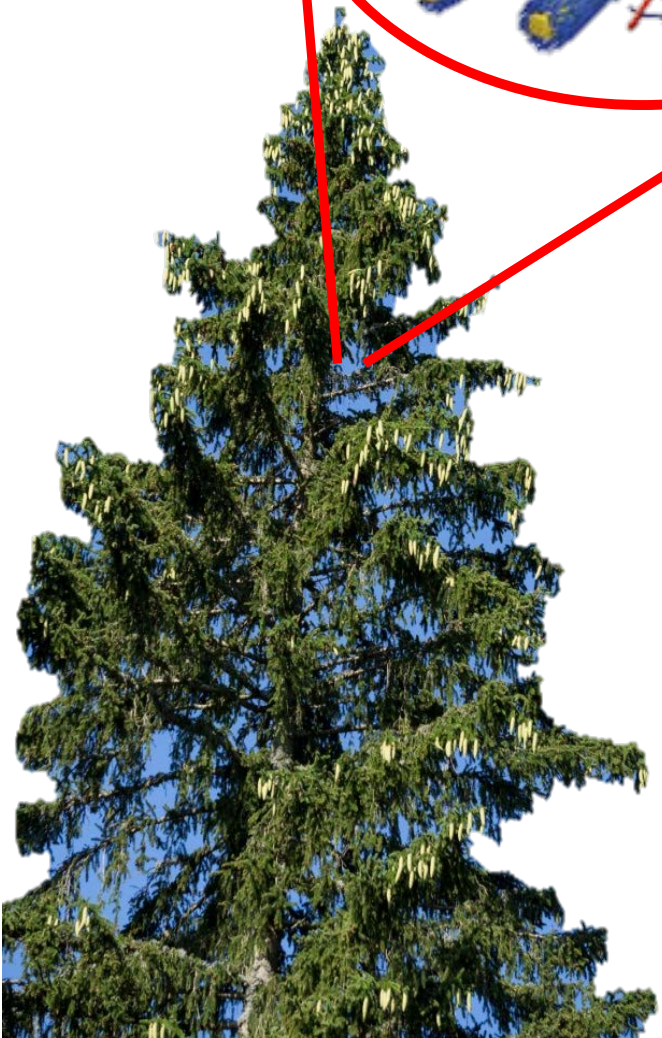


Lignin („3-D“)

# The three main wood components



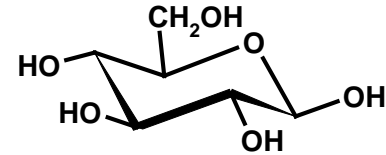
Hemicelluloses („2-D“)



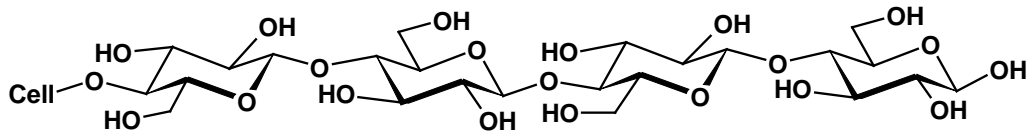


# Cellulose – a short journey through its structure

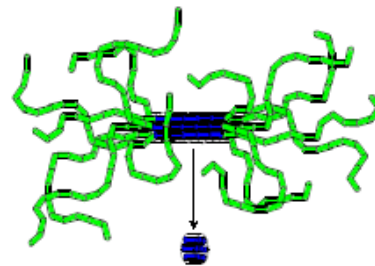
- Homopolymer of D-glucopyranose with  $\beta$ -1-4-glycosidic linkages



- Linear cellulose molecules are stiffened by intramolecular H-bonds

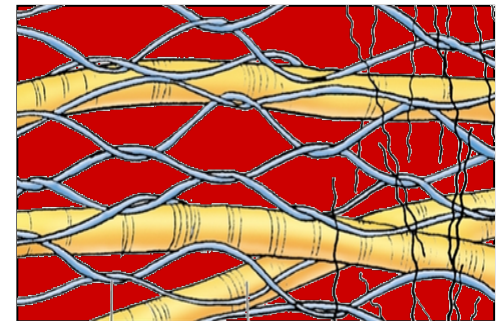
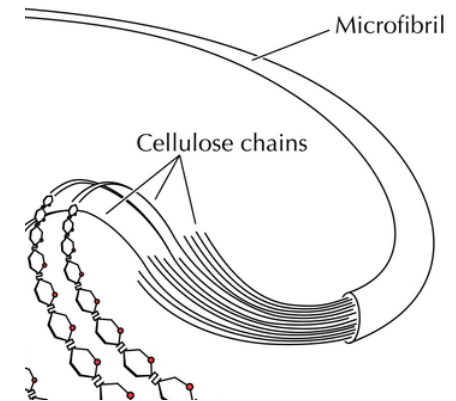


- Intermolecular H-bonds between the chains:  
Formation of microfibrils

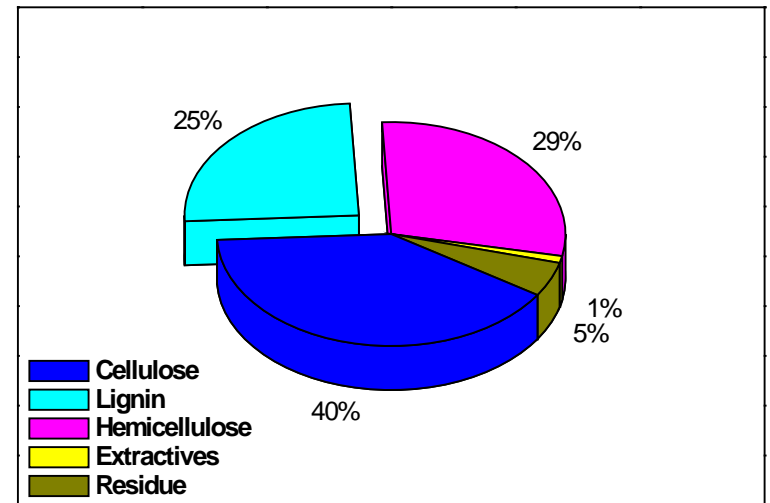


- According to the degree of order:  
crystalline and amorphous regions

- Macrofibrils in the layers of the cell wall in different orientation, lignin and hemicelluloses connect and „glue“ the network of fibrils into a high-performance natural composite material: wood



# Lignin

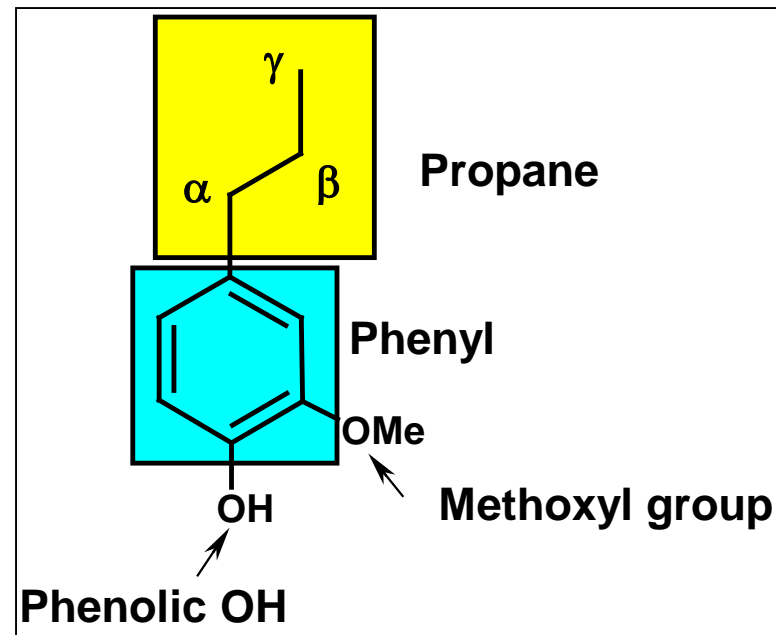


- **Lignin** accounts for approx. 30% of organic carbon in the biosphere. It is the second most abundant natural polymer next to cellulose.
- Lignin was essential in the evolutionary adaptation of plants from aquatic environments to land.
- Lignin waterproofs the cell wall, enabling transport of water and solutes through the vascular system.
- Lignin plays a crucial role protecting plants against pathogens.
- Lignin is responsible for the mechanical strength and stiffness of plants and - most importantly - of trees.

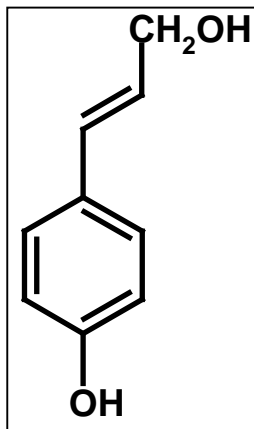


# Lignin

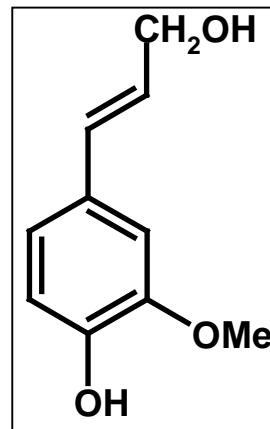
## Phenylpropane units



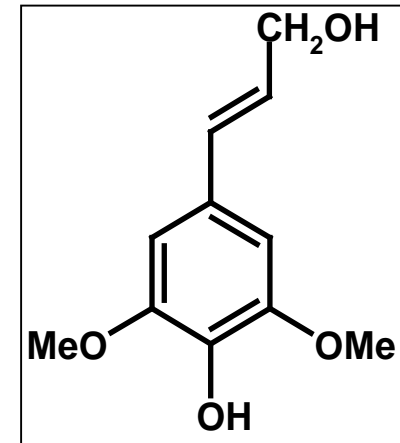
- Lignin consists of **three phenylpropane units**, connected by many **different linkages** formed in a random radical polymerization process.



*p*-Cumaryl alcohol

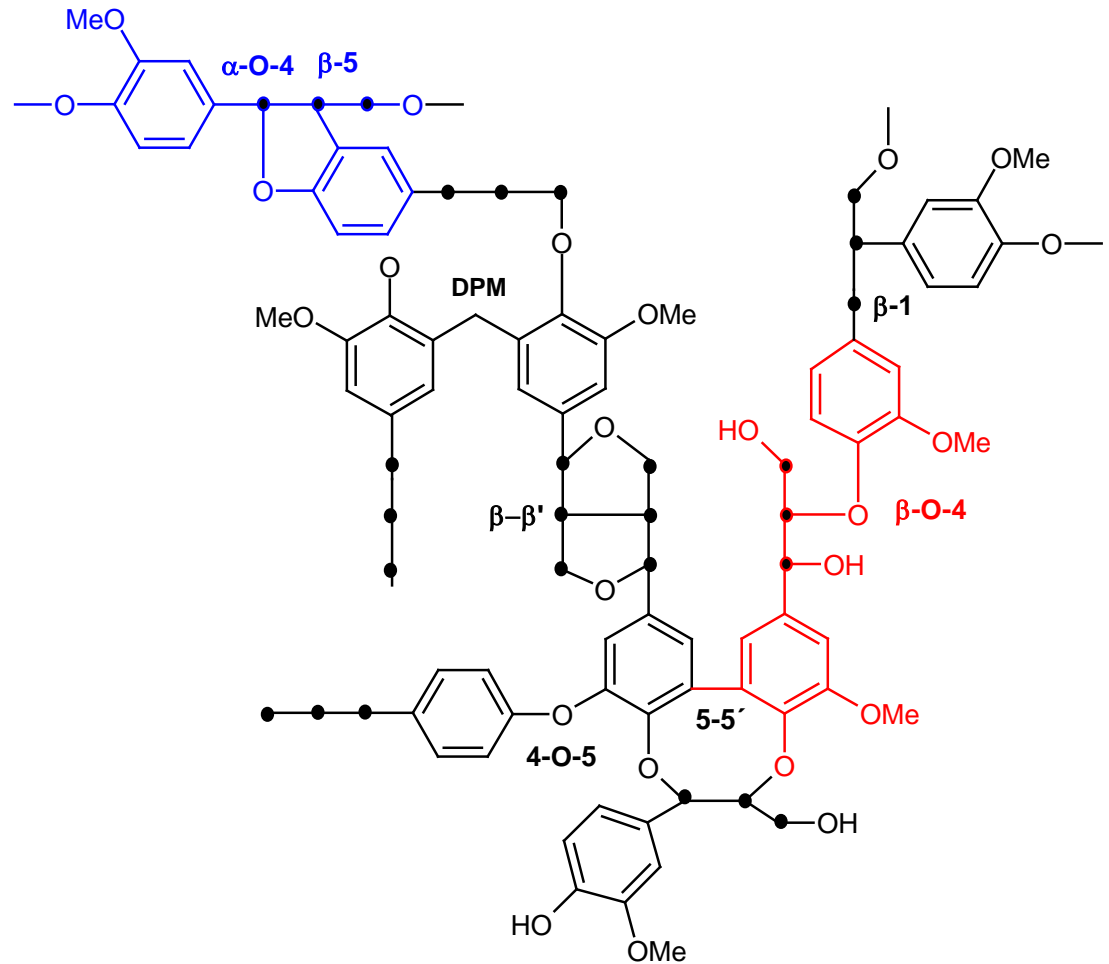


Coniferyl alcohol



Sinapyl alcohol

# Lignin

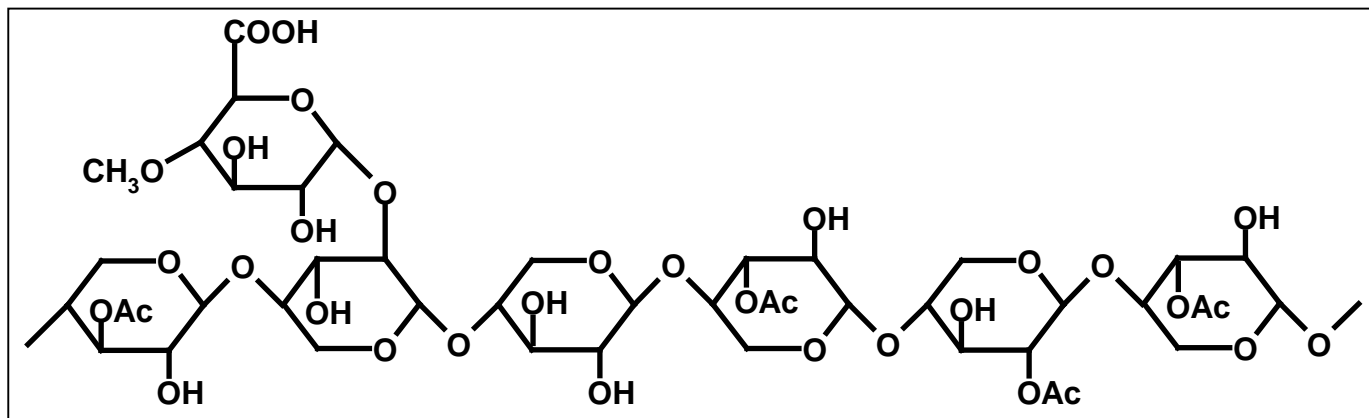


- Lignin consists of **differently linked phenylpropane units**.
- Lignin is an **irregular polymer** and a **polyphenol**.
- Lignins are **complex, racemic, aromatic, yellow-to-brown, hydrophobic heteropolymers**.

# Hemicelluloses



- contain **different sugar “monomers”** by contrast to cellulose (hexoses, pentoses, glucuronic acid, deoxy-hexoses) → **heteropolymers**
- much shorter chain than cellulose, lower molecular weight
- **branching**, short side chains, partly acetylated
- act as “compatibilizer” between hydrophilic cellulose and hydrophobic lignin
  
- **Hardwood hemicelluloses** contain predominantly **xylose units (xylans)**,
- **Softwood hemicelluloses** contain predominantly **mannose units (mannans)**.



Structural detail from hardwood xylan: O-acetyl-(4-O-methylglucurono)xylan



# Comparison Comparison



- carbohydrate (“1D”)
- one monomer
- one linkage
- highly regular, linear
- H-bond network
- species-independent
- white
- hydrophilic
- utilization long established
- so far the main target of pulp&paper industries

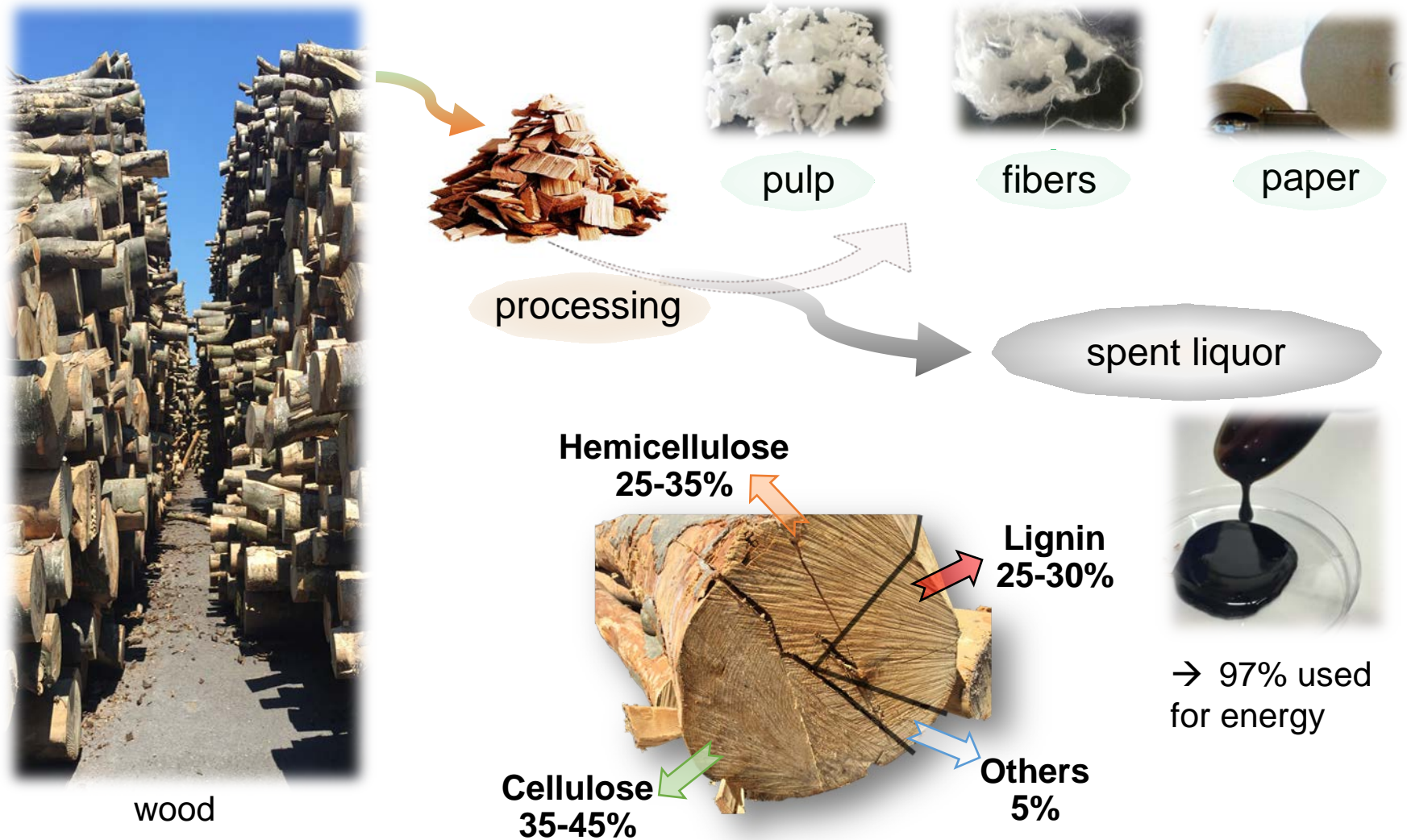


- aromatic / aliphatic
- three monomers
- many different linkages
- highly irregular (“3D”)
- species-dependent
- yellow-brownish
- hydrophobic
- utilization long neglected
- becomes the main target of biorefineries (C-source)

- carbohydrate (“2D”)
- several monomers
- few different linkages
- largely regular, branched
- species-dependent
- white
- hydrophilic / amphiphilic
- utilization long neglected (remained “hidden” in paper or lignin)

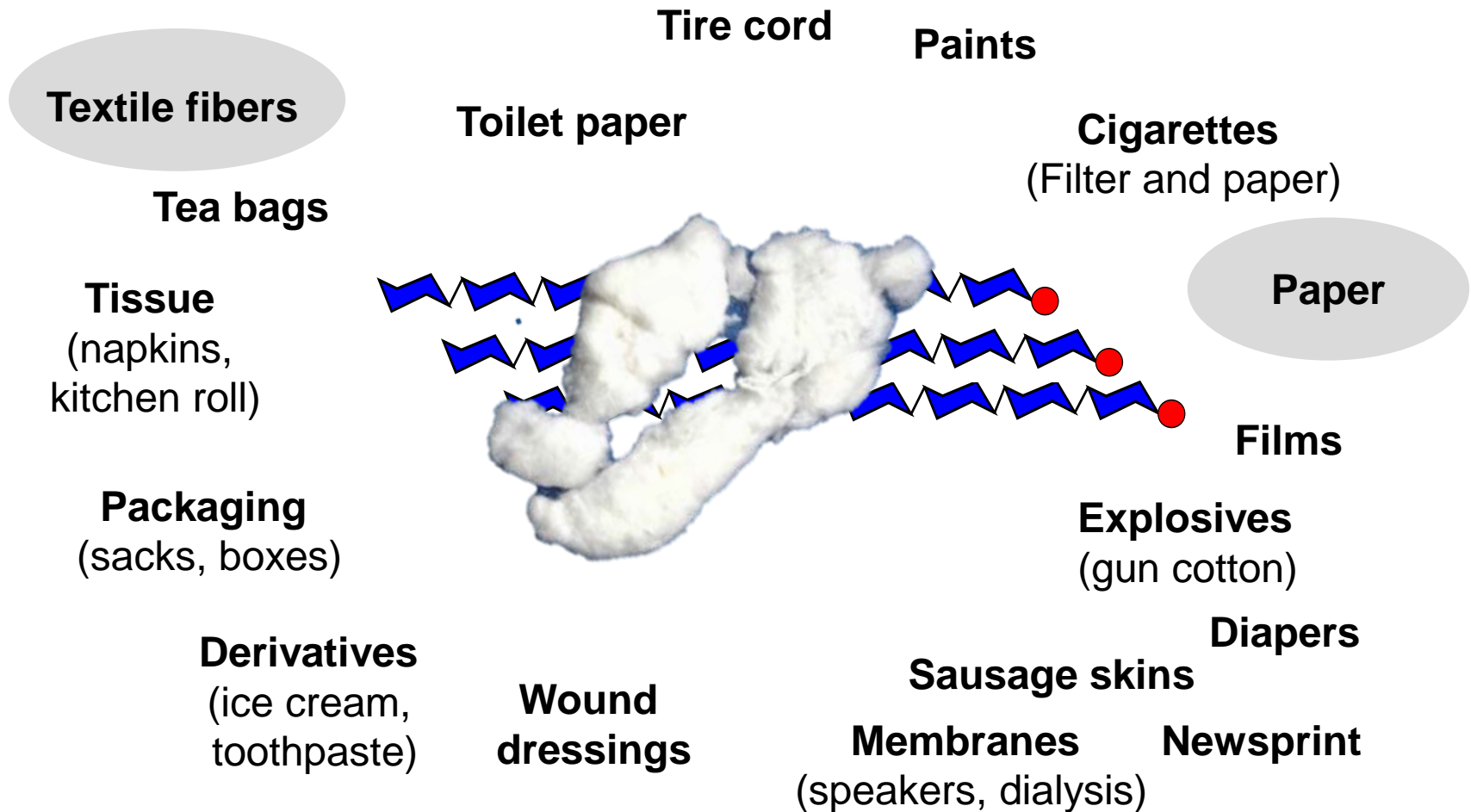


# Chemical composition and usage of wood



**Annual production of pulp: ~ 400 Mtons**

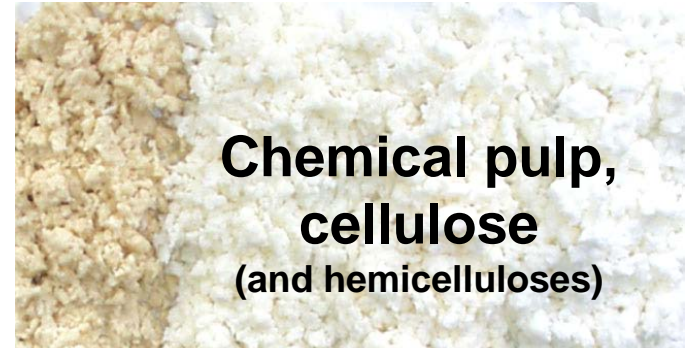
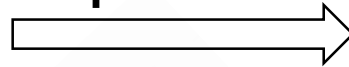
# We know quite well what to do with cellulose...



Wood / → Lignocellulose → Cellulose (pulp) + “black liquor”  
Annual Plants



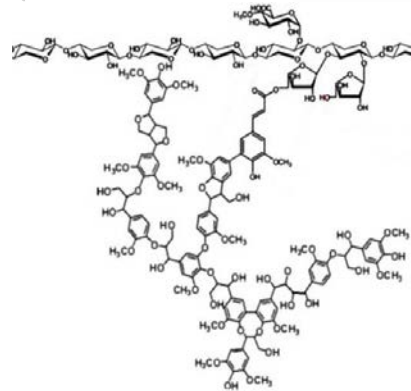
Pulping  
process



Chemical pulp,  
cellulose  
(and hemicelluloses)



**LIGNIN**  
20-30% in  
woody biomass

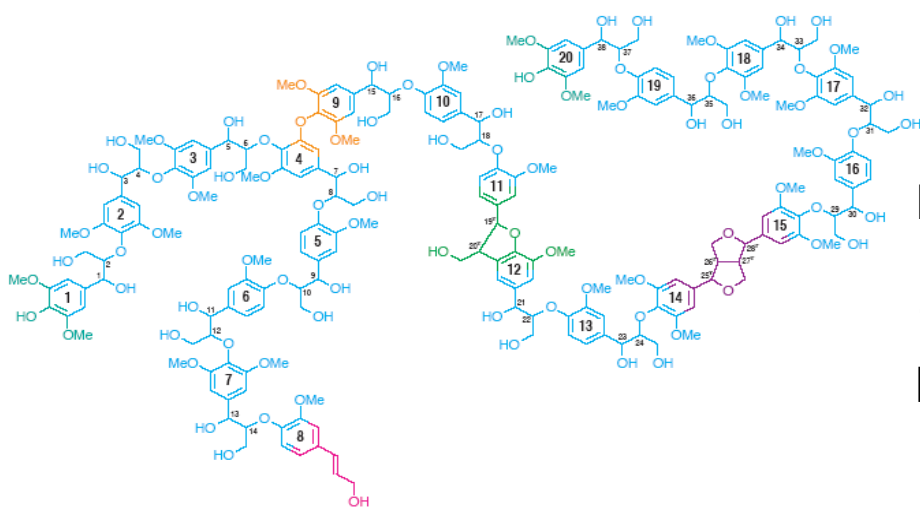


**Black Liquor**  
with dissolved  
**lignin and hemicelluloses**  
(mixture of organics and inorganics)





# The fates of cellulose and lignin...



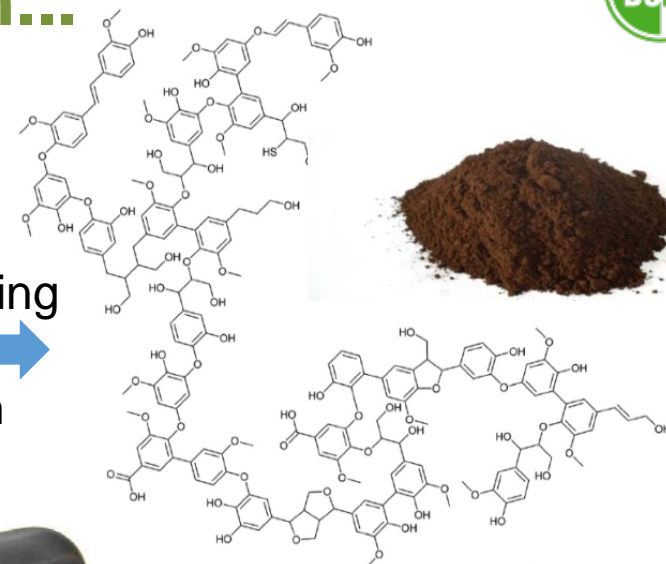
**Native Lignin**



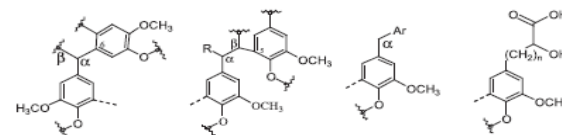
Kraft pulping



170°C, 4h  
pH 14



**Technical lignin**

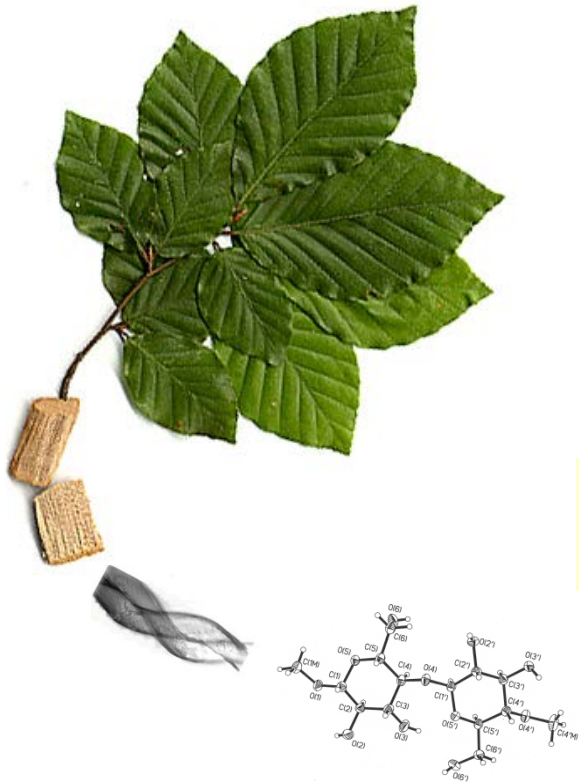


Technical lignins have structures largely different from those of native lignins.





# Wood as starting material in biorefineries



Wood and its constituents -  
in nature

Wood and its constituents -  
in biorefineries



Six problems of today's  
biorefineries and green chemistry  
(„Sixpack problems“)

Brief conclusion



# The sixpack of biorefinery problems in bioeconomies

## 1. The “time” problem

→ Time to grow: refinery 130+ years, biorefinery 25+ years only

## 2. The “energy vs. matter” problem

→ We need “CARBON” to produce materials and chemicals, but not necessarily for energy production!

## 3. The “lignin” problem

→ Cellulose (and hemicellulose) can be nicely used today, lignin not yet.

## 4. The „separation / analysis“ problem

→ Demand for new biorefinery-specific separation / purification / analysis techniques !!!

## 5. The “alpha – beta” problem

→ “Glucose from starch” is easy, “glucose from cellulose” is a lot more difficult.



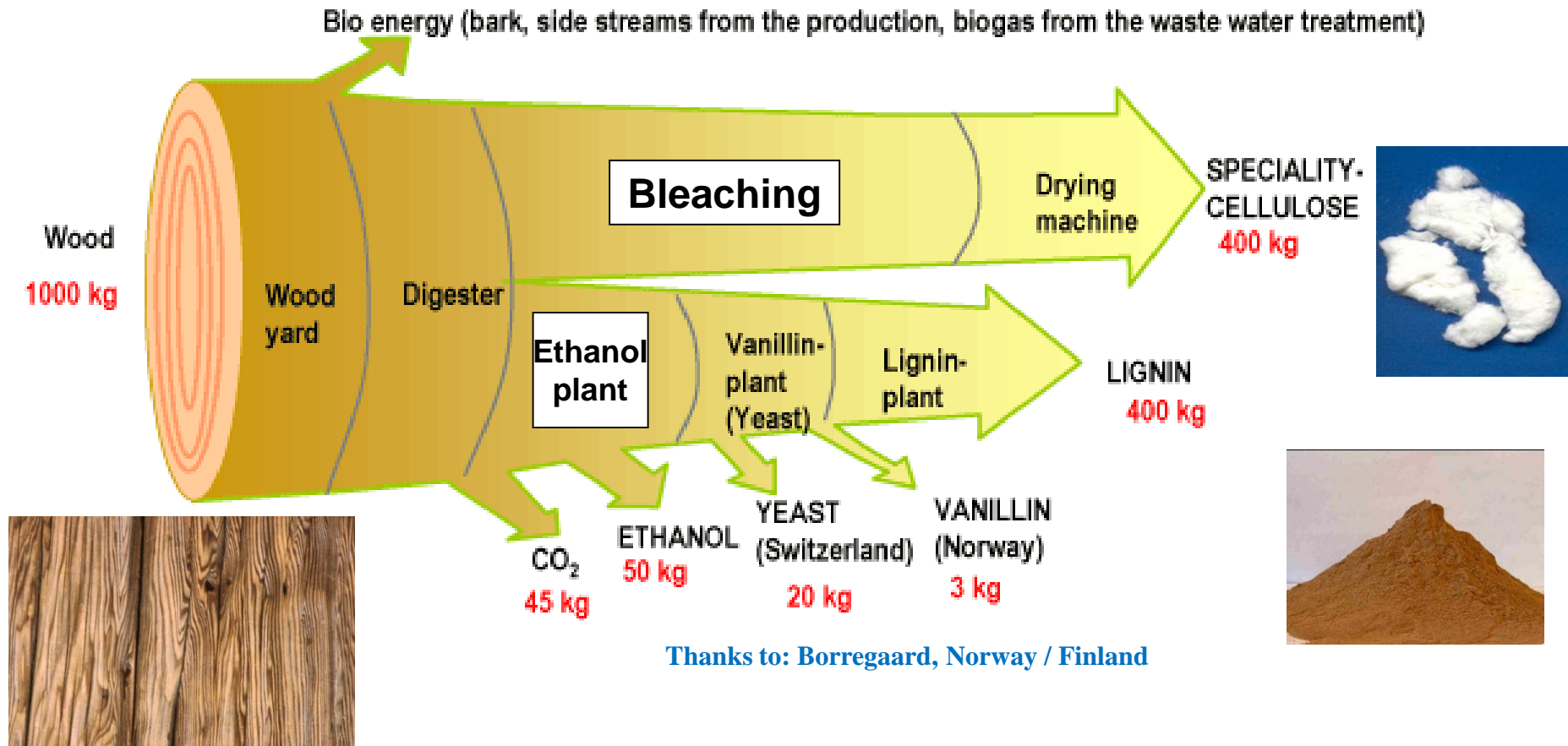
## 6. The “breakdown” problem

→ Is it wise to largely disregard the synthesis and optimization effort of nature?



# 3. The “lignin problem”

Example: utilization of spruce wood



→ Cellulose (and hemicellulose) can be nicely used today, lignin not yet.

→ **Meaningful** uses of lignin as a general C-source for the chemical industries are needed !!!



## 4. The „separation / analysis problem“

### Natural products

Extremely complex mixtures

Natural variability

Hard to process (consistency)

Unknown components

Mostly aqueous mixtures

Often low concentration  
(fermentation)

Unstable upon storage

→ Demand for new biorefinery-specific  
separation / purification /  
analysis techniques !!!



# Lignin analysis (and most utilization modes) start with lignin isolation from black liquor !!

Technical lignin  $\neq$  technical lignin.

Kraft lignin



Lignosulfonate



Organosolv



Flüppr<sup>o</sup>

*Lignin platform*

A typical biorefinery matrix problem:

too  
much  
salt

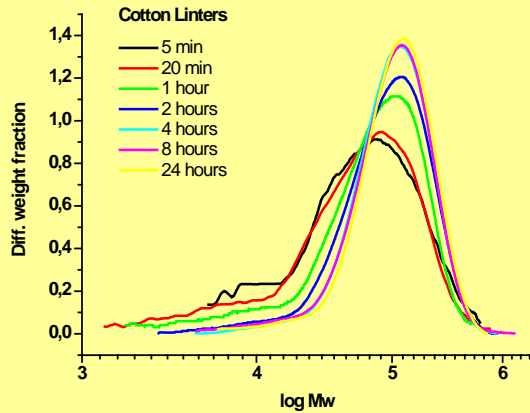




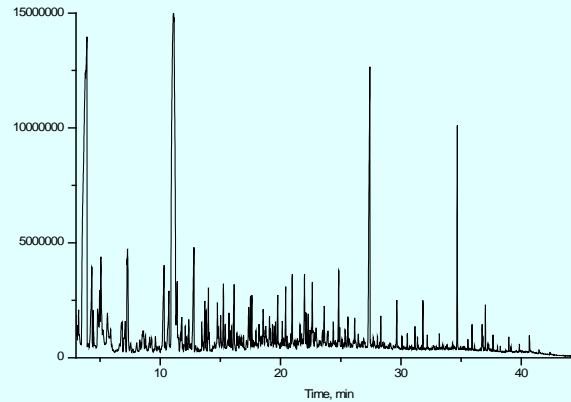
# We need powerful „Biorefinery analytics“



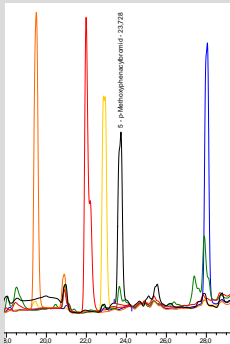
## Carbohydrate (cellulose) analysis



## Lignin analysis

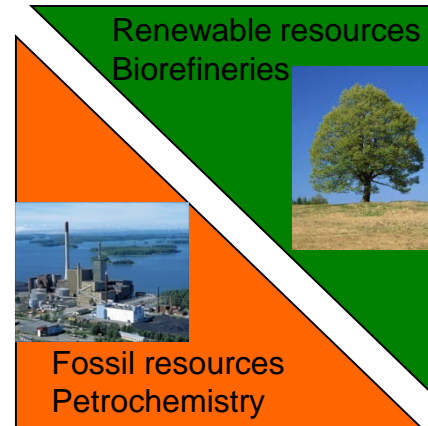


## Analysis of byproducts



- GC-MS
- CE-MS
- NMR
- LC-MS
- DESI-MS
- GPC

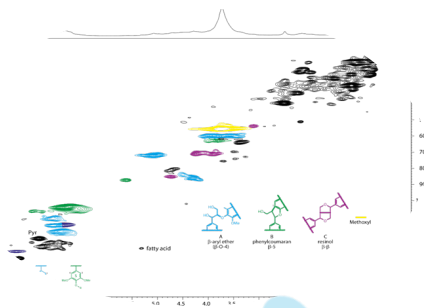
**Development of new products and technologies based on renewables must go hand in hand with the development of robust and reliable analytical methodology.**



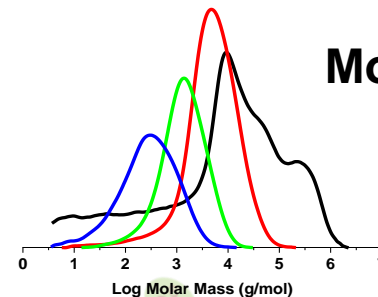
# Comprehensive analysis of technical lignins

NMR

(QQ) 2D,  $^{31}\text{P}$ ,  $^1\text{H}$ ,  $^{13}\text{C}$



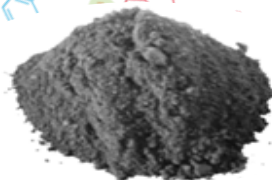
Molar mass



Nitrobenzene oxidation  
 $\text{KMnO}_4$  oxidation

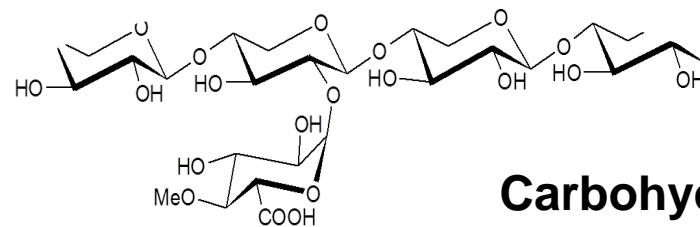
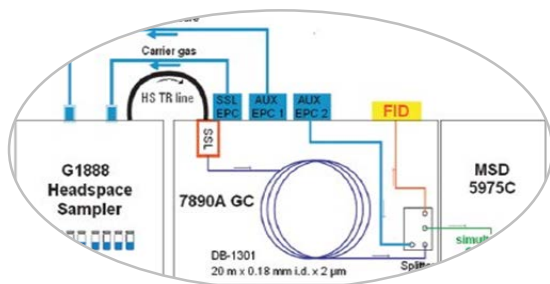
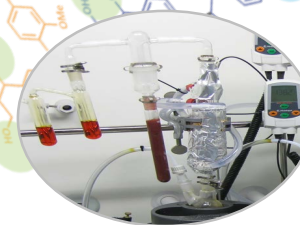
Hydroxyls,  $\text{COOH}$ ,  
 $-\text{SO}_3\text{H}$ , ...

Technical lignins



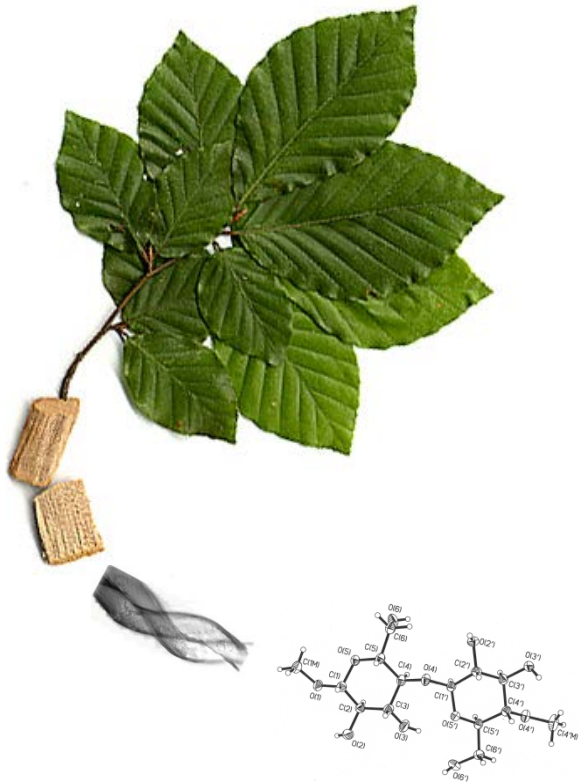
DFRC  
 Thioacidolysis

$-\text{OMe}/\text{OEt}$



Carbohydrates

# Wood as starting material in biorefineries



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Wood and its constituents -  
in biorefineries

Six problems of today's  
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Brief conclusion



**ABCT** (Austrian Biorefinery Center Tulln) –  
**ABCM** (Doctoral School "Advanced Biorefineries –  
Chemistry & Materials)




Universität für Bodenkultur Wien



All PhD candidates working in the **Austrian Biorefinery Center Tulln (ABCT)** are members of the **ABC&M doctoral school**.

The combination of research and PhD education, realized by the **linkage of ABCT to the doctoral school ABC&M**, ensures the availability of future biorefinery experts and long-term continuity in research.





**ABC&M**

**DOCTORAL SCHOOL**  
**ADVANCED BIOREFINERIES:**  
**CHEMISTRY & MATERIALS**

**Coordination:**  
**Institute of Chemistry of Renewable Resources, BOKU**



# Financial \$upport



*Alma mater viridis*



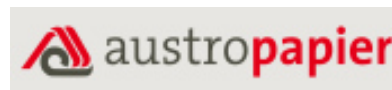
European Polysaccharide  
Network Of Excellence



FWF

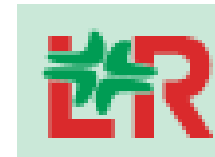


FFG



National Institutes of Health

*The Nation's Medical Research Agency*



kemira

sappi



fzmb



ISOVER



heinzelpulp

ZELLSTOFF PÖLS AG

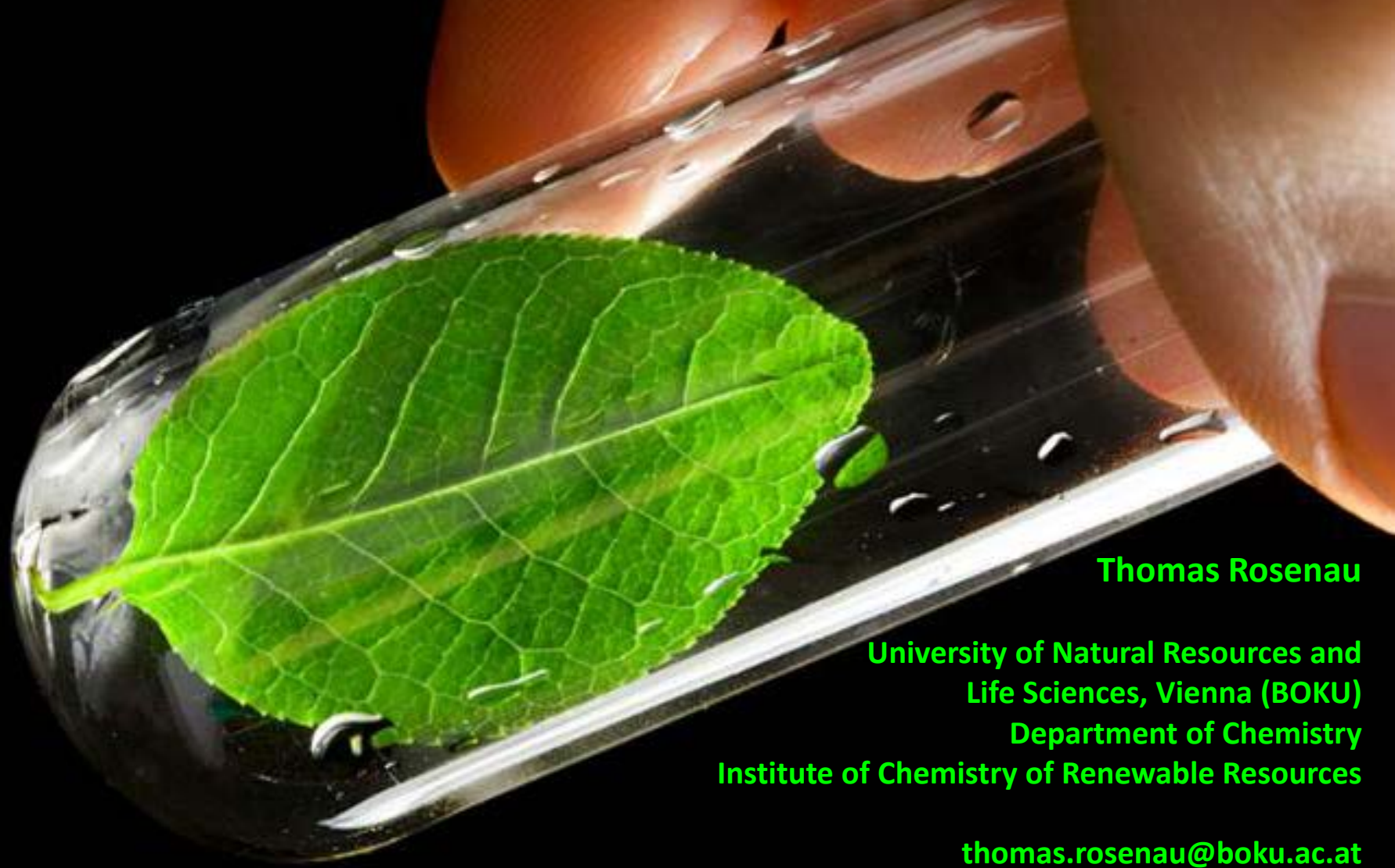
**Thanks to the people who  
are actually doing the work...**





**Thank you for your kind attention !**





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**University of Natural Resources and  
Life Sciences, Vienna (BOKU)  
Department of Chemistry  
Institute of Chemistry of Renewable Resources**

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in biorefineries**